



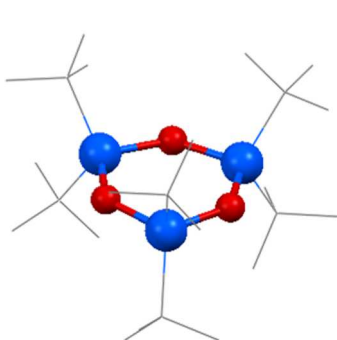
**A NUMERIC MODEL FOR THE IMAGING
MECHANISM OF
METAL OXIDE EUV RESISTS**

W. Hinsberg* and S. Meyers

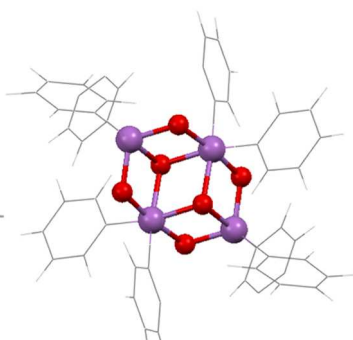
Inpria Corp.

*Columbia Hill Technical Consulting

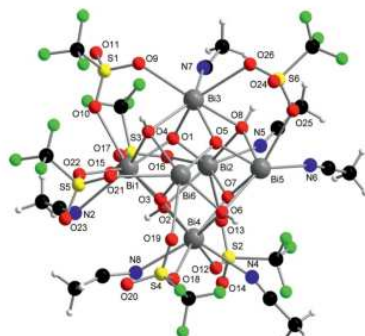
Metal Oxide (MO_x) Cluster Resist Examples



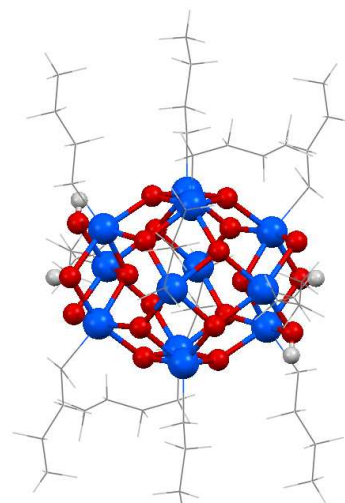
[R₂SnO]₃



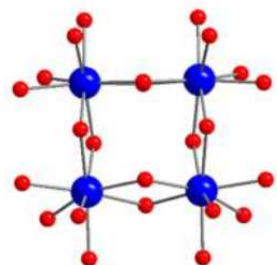
[(R₂Sb)₂O₃]₂



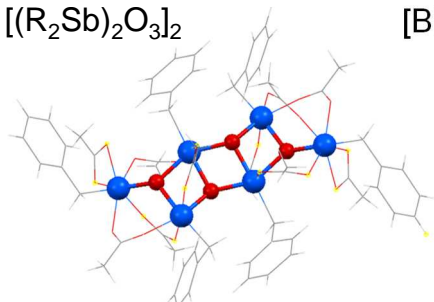
[Bi₆O₄(OH)₄(OTf)₆(CH₃CN)₆]



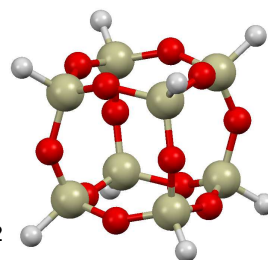
[(RSn)₁₂O₁₄(OH)₆]⁺²



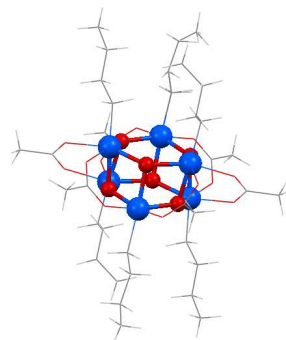
[Hf₄(O₂)(OH)₇(SO₄)₃]⁺



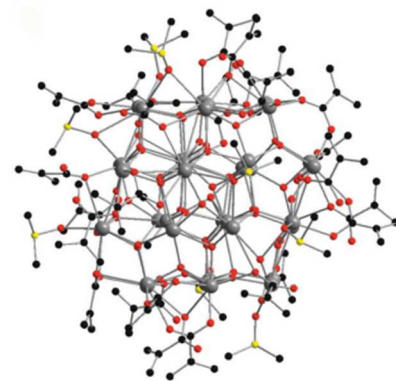
[{RSn(O)OOCR'}₂{RSn(OOCR')₃}]₂



[H₈Si₈O₁₂]

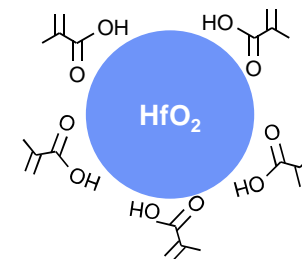
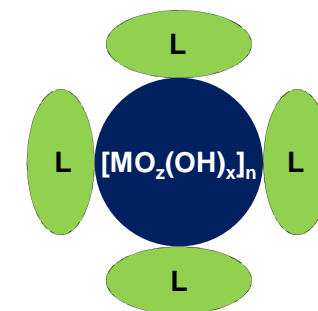


[RSn(O)OOCR']₆



[Bi₃₈O₄₅(OMc)₂₄(DMSO)₉]

● C
● O
● S
● Bi

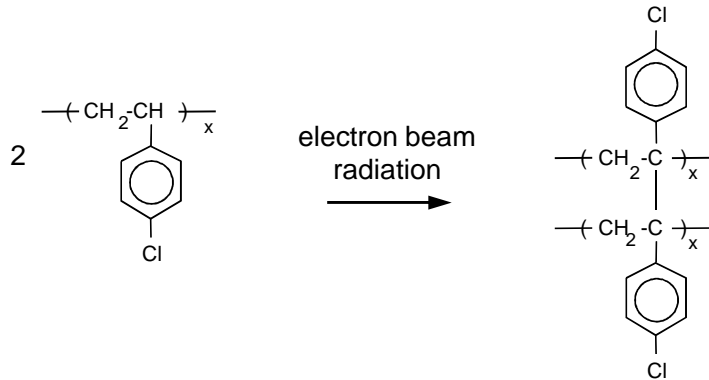


MO₂ Ligand-Stabilized nanoparticle

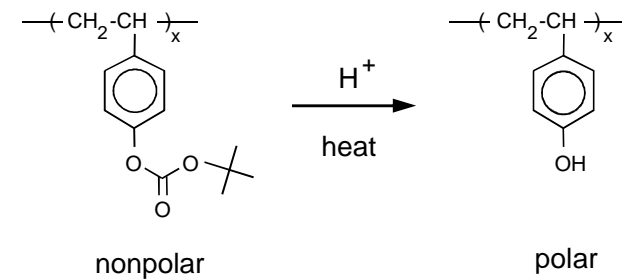
EUVL Workshop 2018 2

Negative Tone Resist Chemistries in Organic Polymers

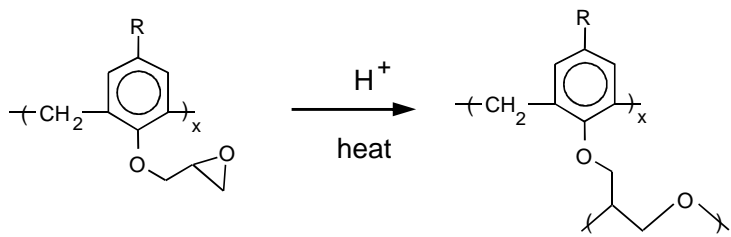
Direct crosslinking of polymer [poly(4-Cl-styrene)]



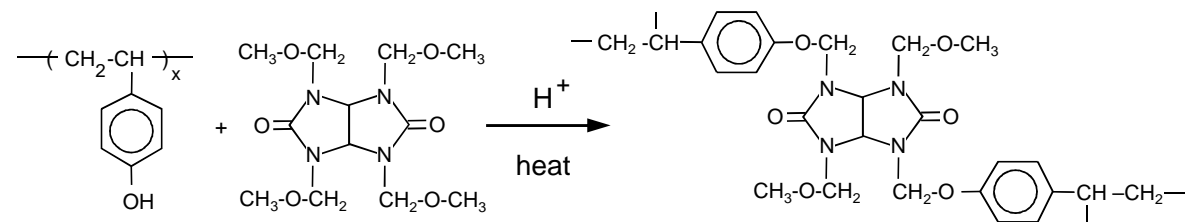
Polarity change [*t*-BOC, NTD resists]



Chain polymerization [Riston, SU-8]



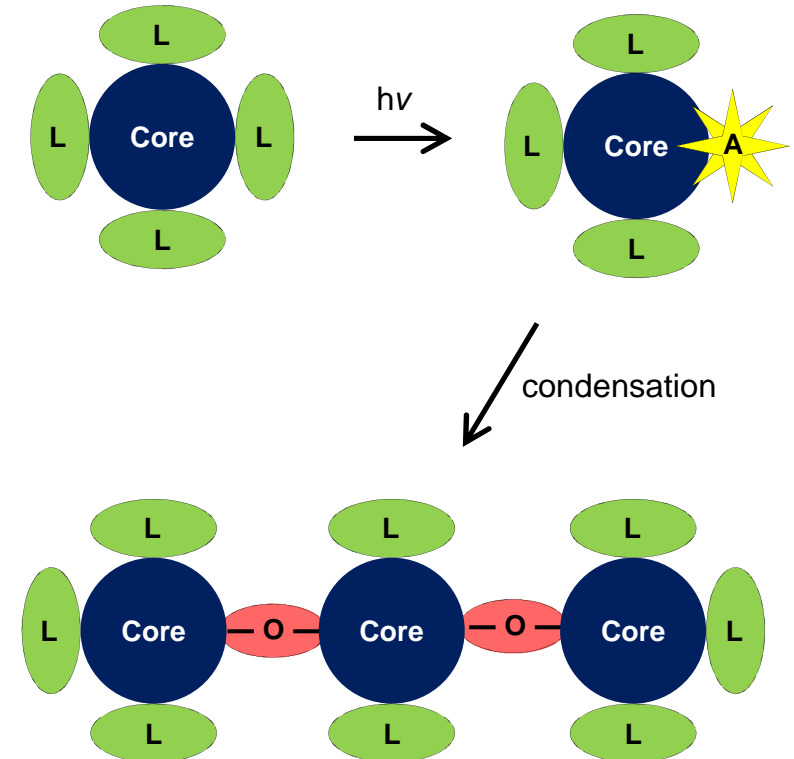
Crosslinking by multifunctional additive [CGR, bis-azide]



None of these describe the primary imaging mechanism in MOx resists

A Generic Imaging Model for MOx Resist Systems

- Basic approach :
 - Minimalist description of chemistry and physics
 - Follow progression of radiation and condensation chemistry
- Resist building block notation
 - **Core** molecule / framework / cluster / nanoparticle
 - Multiple **radiation-sensitive ligands L** per core
 - Condensation of **active site A** formed upon ligand radiolysis leads to **Oxo-Network** formation



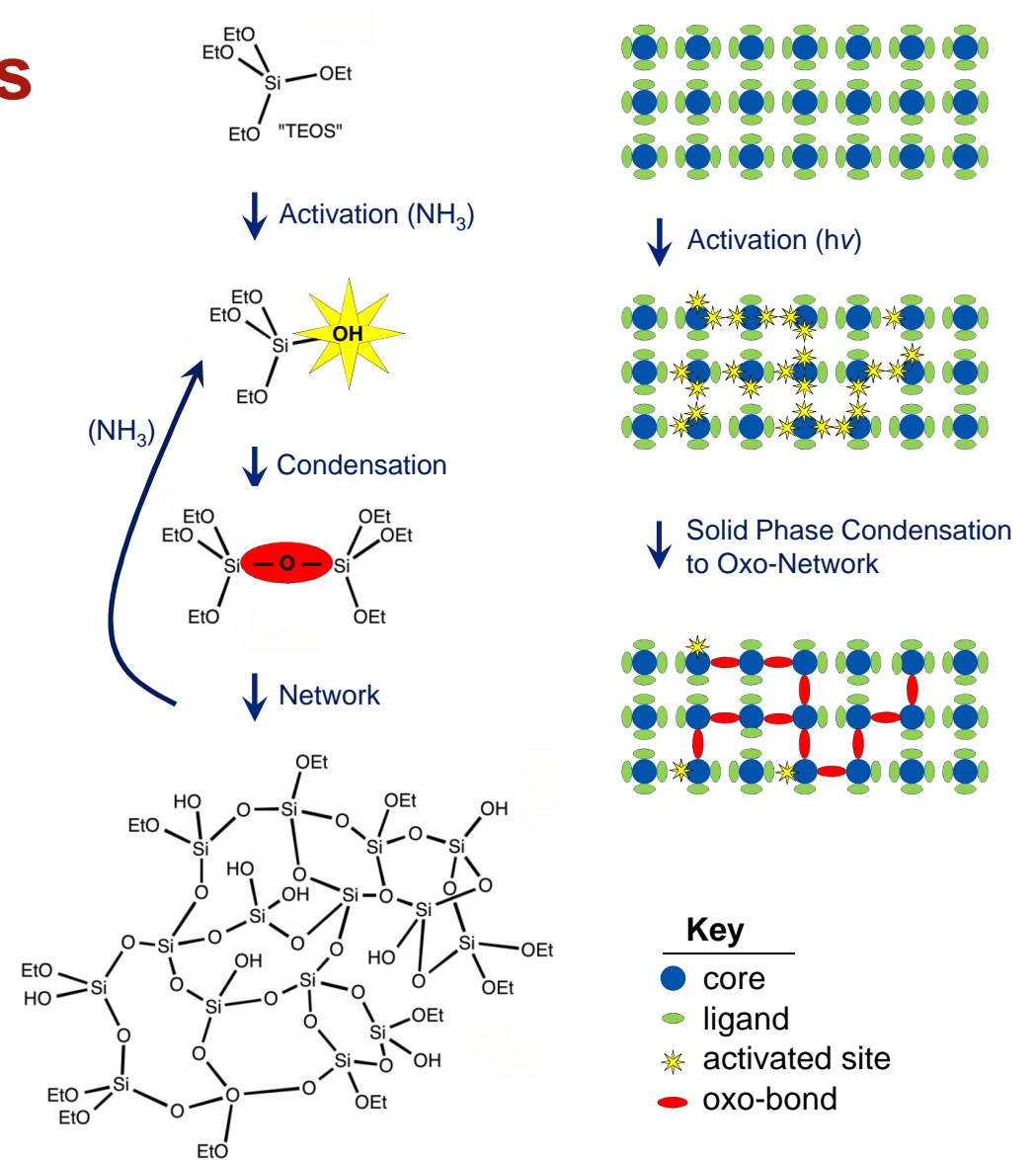
Analogue of Sol-Gel Process

Characteristics of sol-gel chemistry:

- Site activation by catalyst
- Simultaneous condensation and further activation
- Complex mixture of intermediate species
- Polymerization proceeds toward oxo-network formation

MOx imaging model:

- Site activation by radiation chemistry
- Solid phase condensation of neighboring cores
- Complex mixture of intermediate species
- Polymerization proceeds toward oxo-network formation



Overview of Modeling Process

A Molecular-Scale Description

- 3D array of individual molecules
- Track individual events
- Statistical effects accounted for

Model Inputs:

- Molecular volume
- Number of radiation-sensitive ligands per core
- EUV absorption coefficient
- “Quantum yield”:
 - Definition: **number of ligands fragmented per photon absorbed**
 - In model, use the number of electrons generated per photon as a proxy
- “Radiochemical blur length”:
 - Definition: **distance scale over which chemical change may occur from point of photon absorption**
 - In model, use “electron blur” length as a proxy
- Film thickness
- Exposure dose

1. Calculate EUV light absorption

↓ 3D spatial distribution of photon absorption

2. Calculate secondary electron generation

↓ 3D spatial distribution of generated electrons

3. Calculate photoproduct distribution

↓ 3D spatial distributions of each photoproduct

4. Calculate condensation reactions

↓ 3D spatial distribution of oxo-bonds formed

5. Analyze condensation product

↓ 3D topology of oxo-network

6. Imaging properties

Step 1 - EUV Absorption

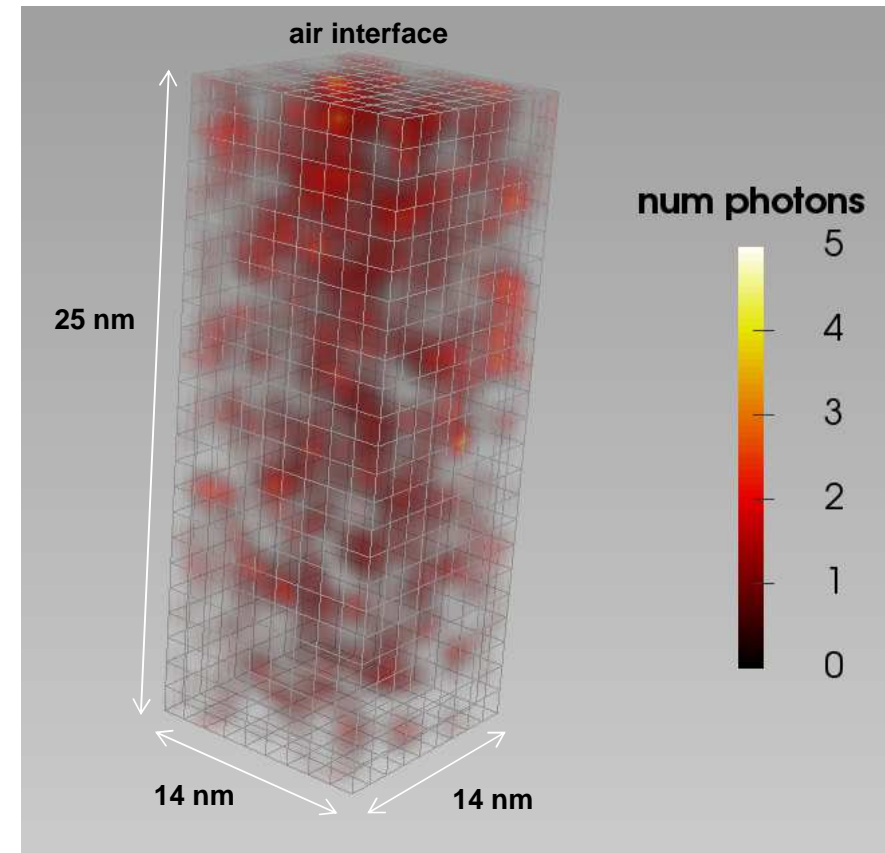
A stochastic implementation of Beer's Law

Protocol:

- Divide film surface into sub-areas
- Distribute impinging photons onto surface
- Distribute photons in sublayers according to probabilities

Example

- molecular volume 2.3 nm^3
- 15 mJ/cm^2 flood exposure
- 1928 impinging photons, 662 absorbed



Step 2 - Secondary Electron Generation

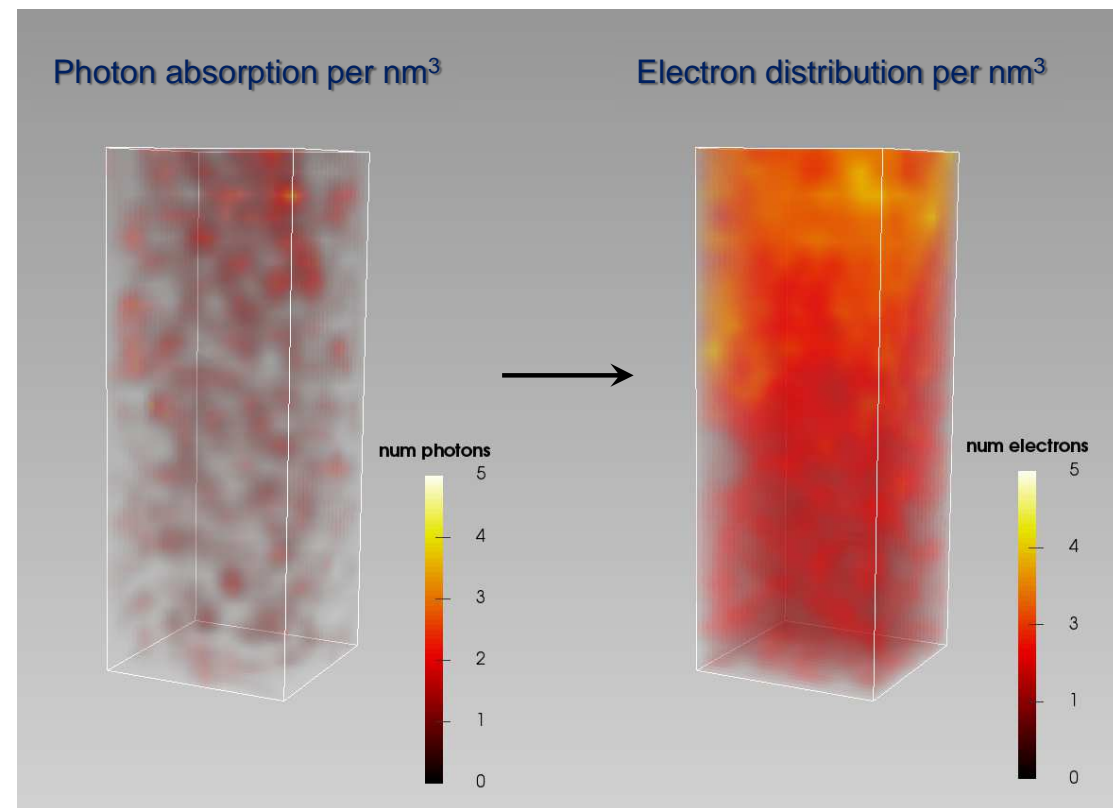
Apply experimentally derived electron yield and blur length to photon absorption distribution

Protocol:

- Each photon generates on average n electrons
- Allow electrons to stochastically “diffuse” from point of photon absorption
- Terminate electron diffusion when average diffusion distance = blur length

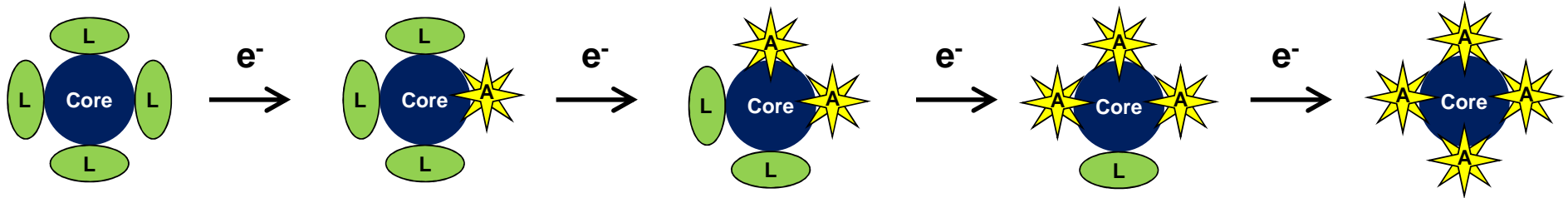
Example

- molecular volume 2.3 nm^3
- 15 mJ/cm^2 flood exposure
- avg 8 electrons per photon
- 1.4 nm electron blur length



Step 3 – Primary Photoproduct Distribution

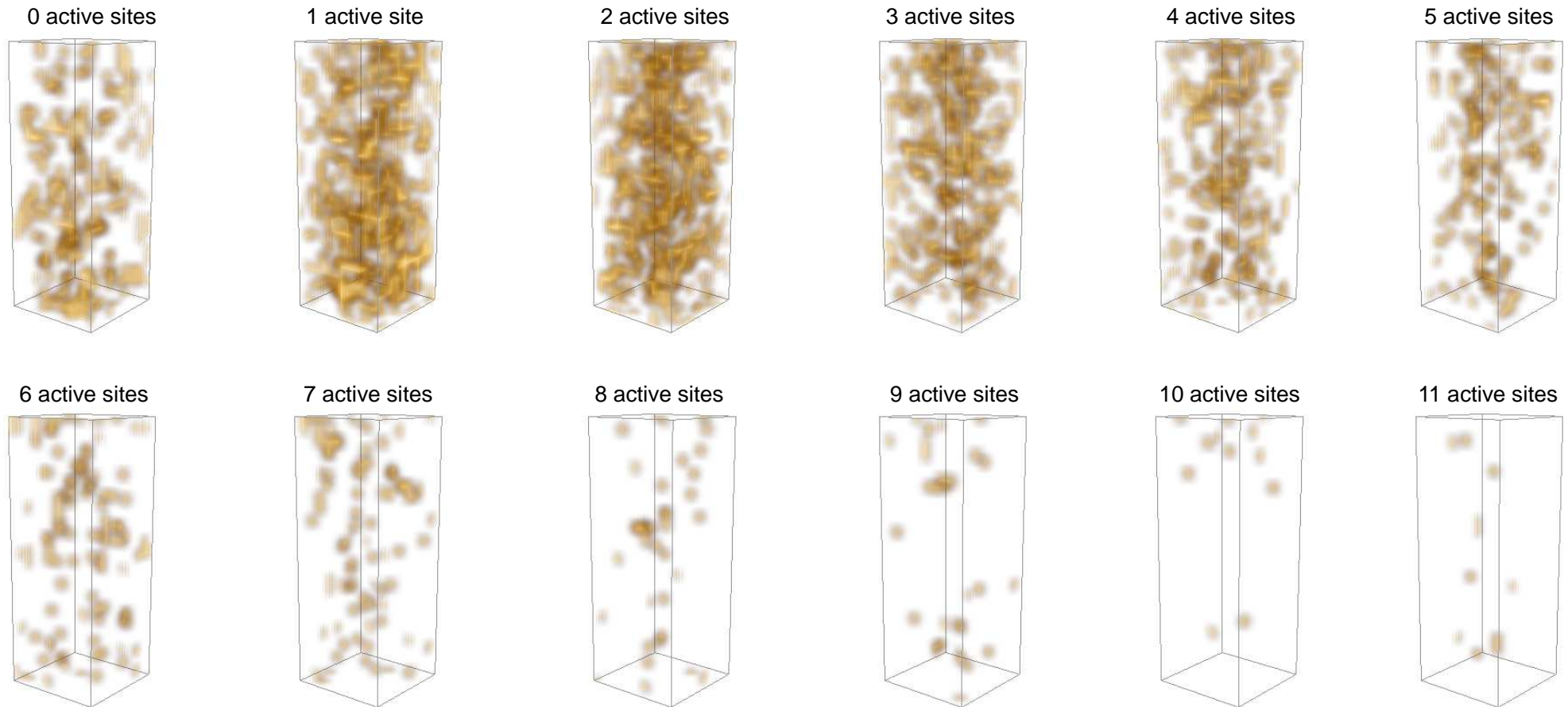
- Allow electron distribution to interact with resist material
- Assumptions:
 - Each ligand is chemically equivalent
 - Reactivity is unaffected by degree of decomposition
- A complex product mix results: e.g., for a tetra-substituted core



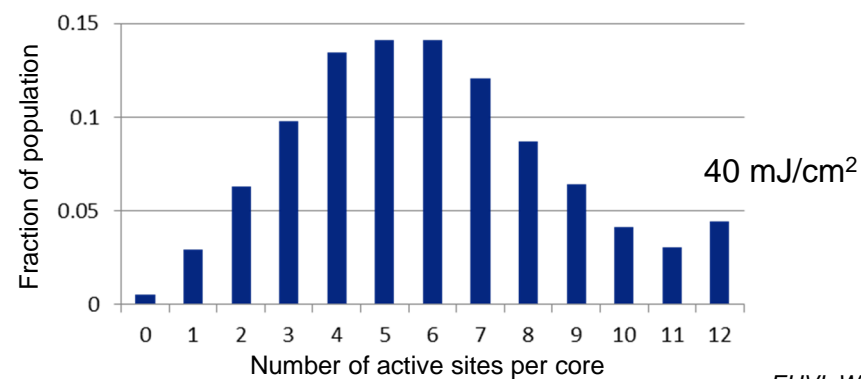
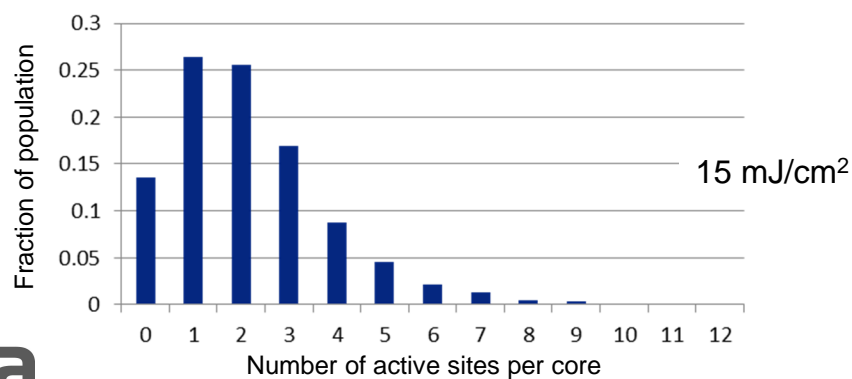
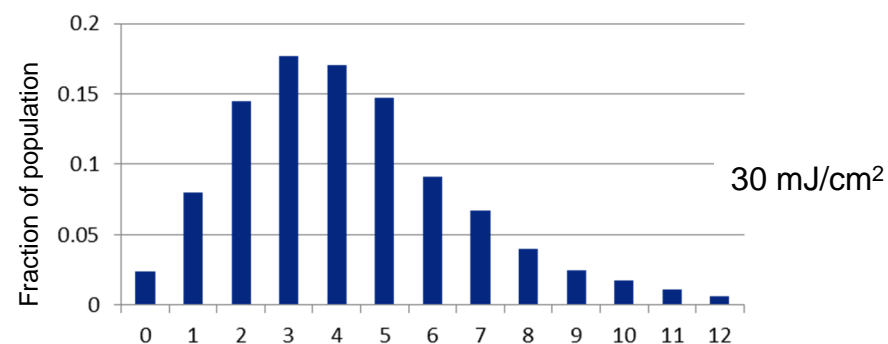
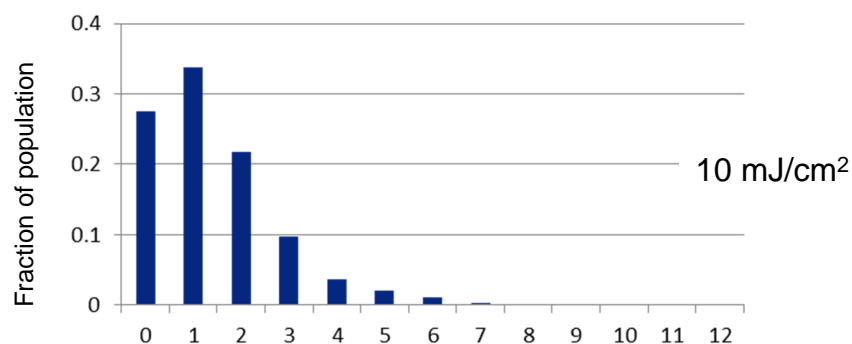
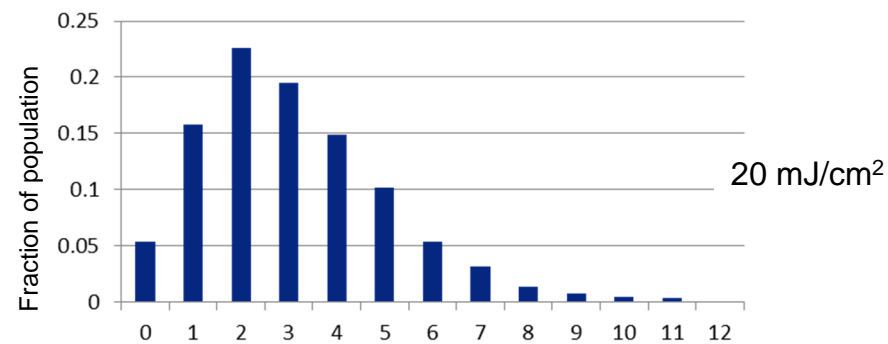
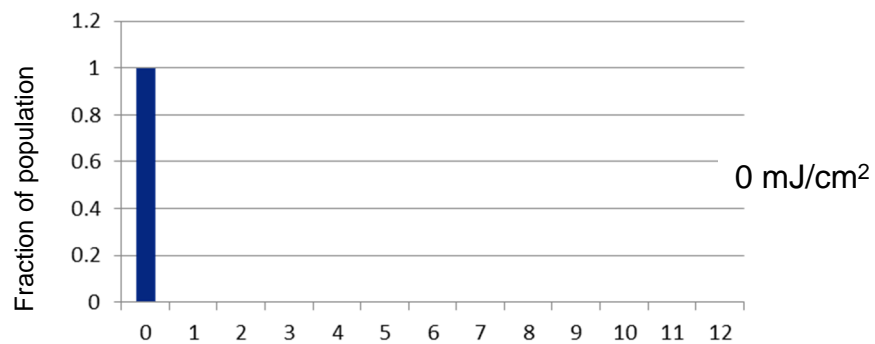
- Photoproduct distribution is a function of dose

An Example Photoproduct Spatial Distribution

Hypothetical MOx resist with 12 radiation-sensitive ligands per core, EUV flood exposure 15 mJ/cm² dose



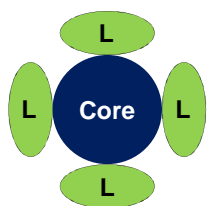
Photoproduct Distribution Depends on Dose



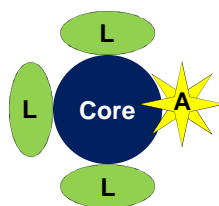
Impact of Photoproduct Distribution

Role of photoproduct in condensation depends on structure:

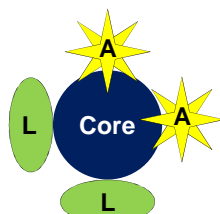
Probability of condensation increases



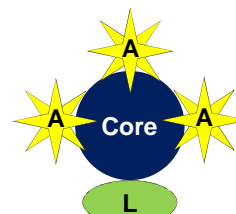
- Inert



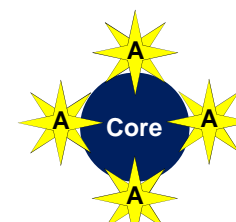
- Terminates chain



- Extends chain



- Extends chain
- Branch point



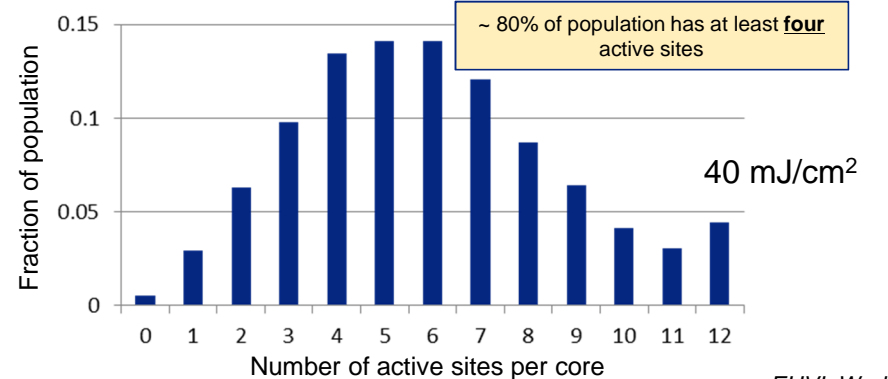
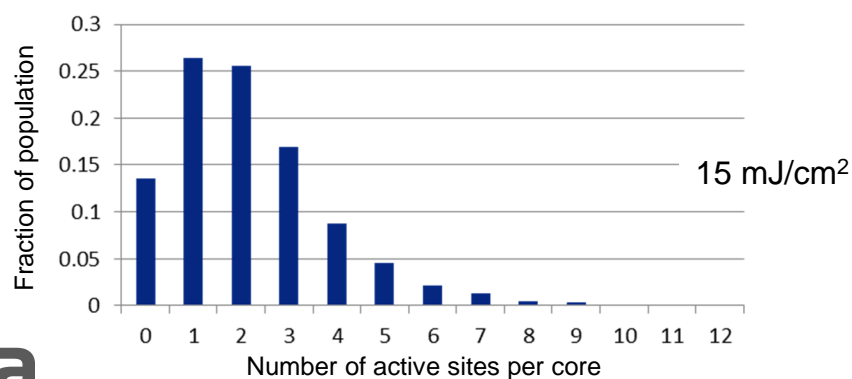
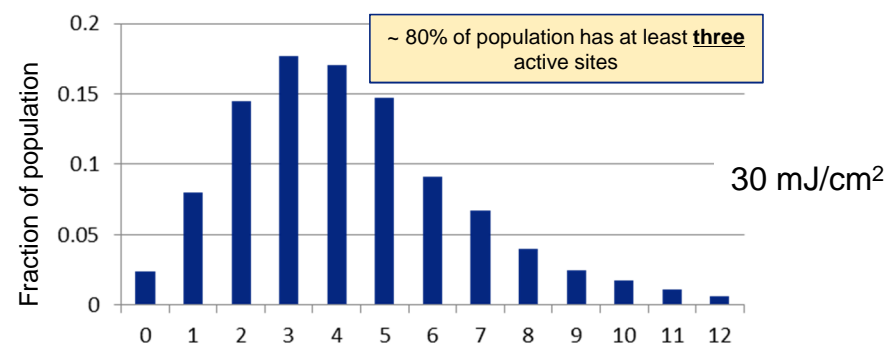
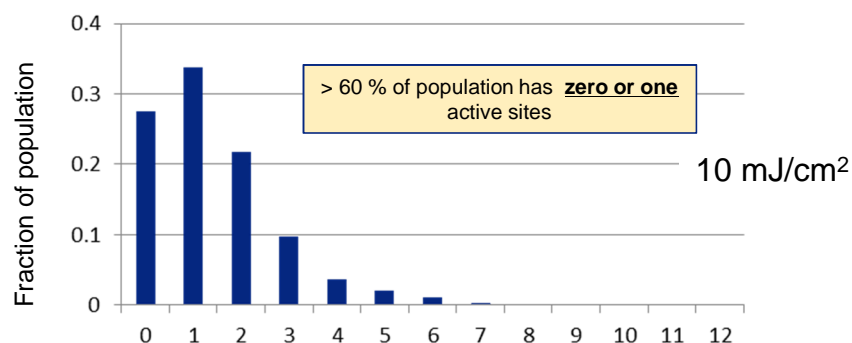
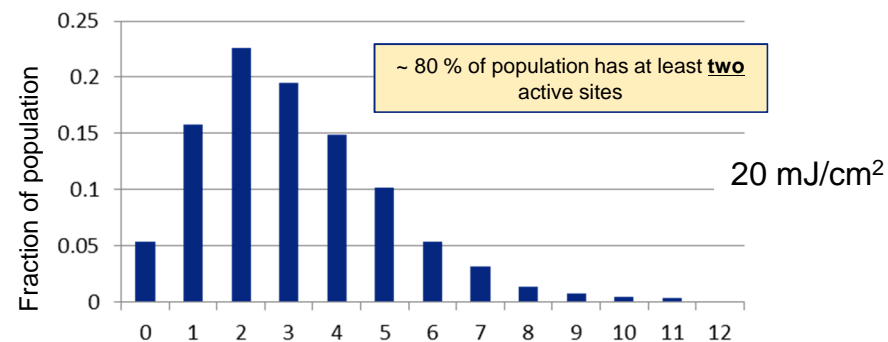
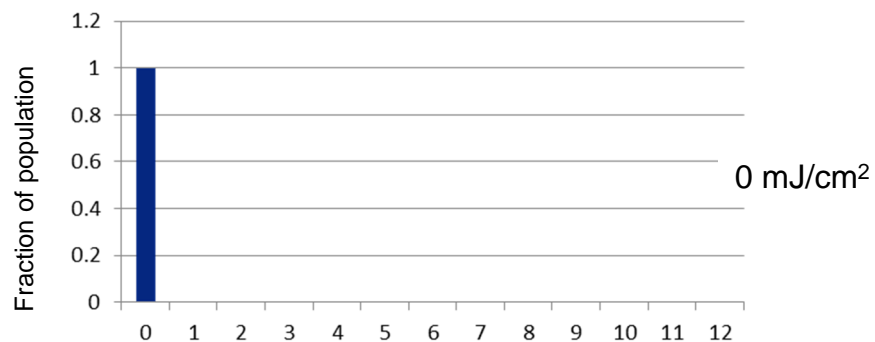
- Extends chain
- Branch point
- Crosslink point

...



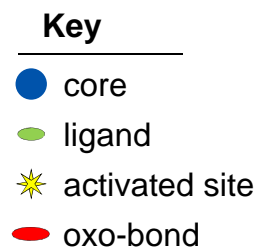
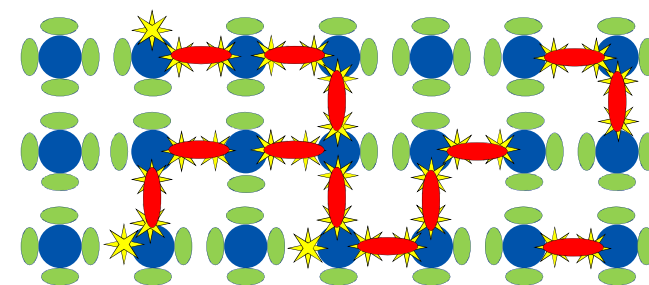
Supports oxo-network formation

Photoproduct Distribution Depends on Dose



Step 4 – Condensation of Primary Photoproducts

- Calculate the evolution of oxo-networks
 - Each core starts with a set number of active sites
 - Condensation only if core and neighbor both have an active site
 - Condensation forms one oxo-bond and consumes two active sites
- Protocol:
 - a) Initialize with photoproduct distribution
 - b) Select a core
 - c) Core and neighbor both have active sites?
 - i. Form oxo-bond
 - ii. Subtract active sites
 - d) Repeat (b) and (c) until probability of reaction is zero

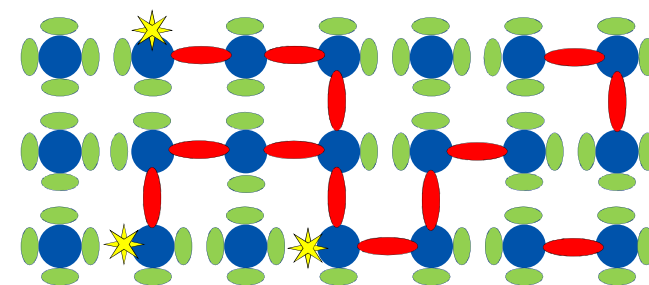


Result: Population of oxo-network polymers

Step 5 – Analyze Topology of Condensation Products

■ Protocol:

- a) Scan through array of cores looking for bonds
- b) If a core is bonded to any of its neighbors
 1. Check each bonded neighbor to see if it is bonded to its neighbors
 2. Recursively following the bonding including branches and crosslinks
- c) Continue scan until every core has been visited

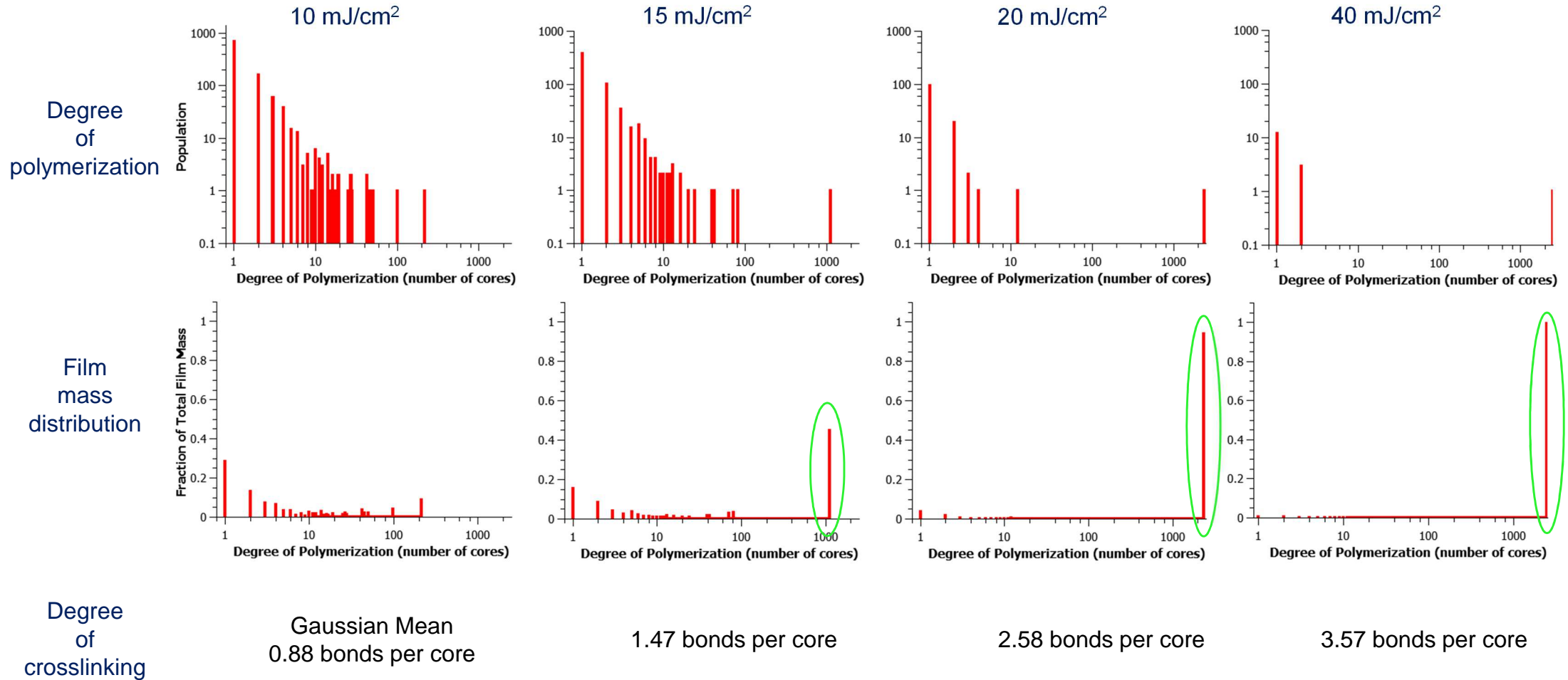


Key

- core
- ligand
- ★ activated site
- oxo-bond

Result: 3D map and population distribution of oxo-network polymers

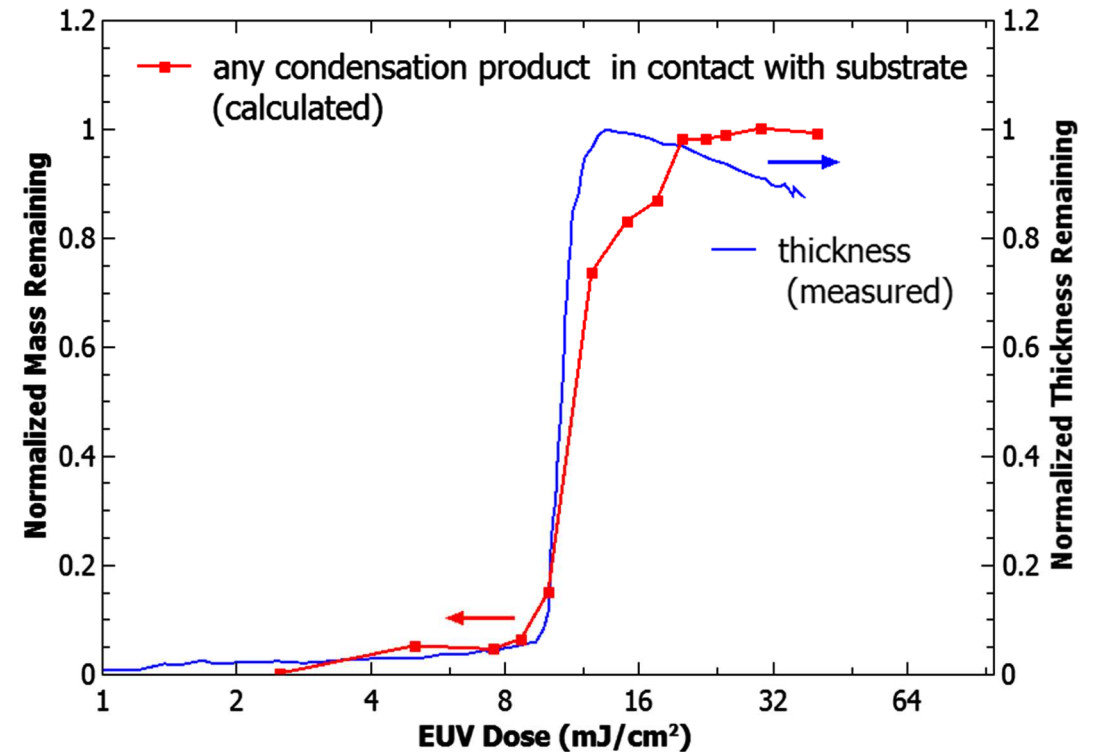
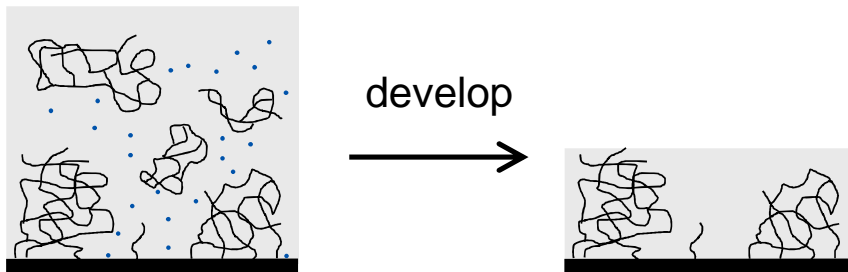
Condensation Product Distribution vs Dose



Calculate Resist Contrast for a Real Resist

Inpria experimental MOx resist system

1. Estimate quantum yield and blur length from experimental data
2. Use model to calculate condensation products vs dose
3. Apply binary dissolution process
 - Only condensation products in direct contact with the substrate are insoluble

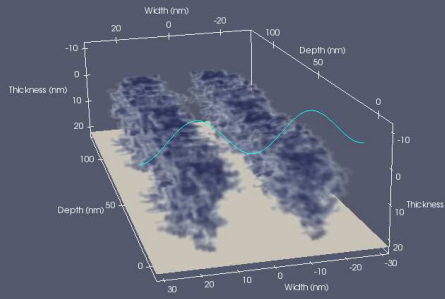


Calculated Line-Space Images

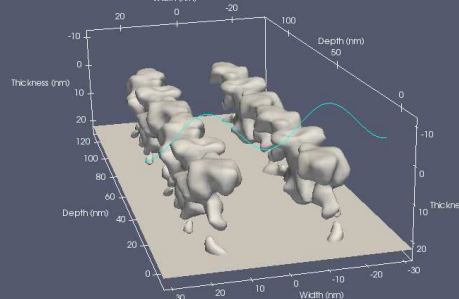
NXE 3300, Dipole illumination, 16 nm line/32 nm pitch

20 mJ/cm²

Map of Oxo-Networks

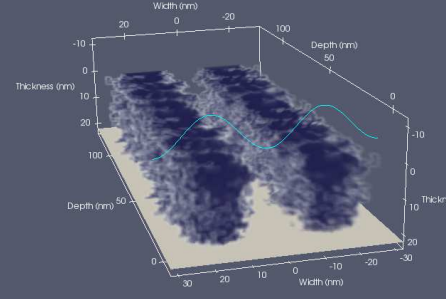


Developed Relief Image

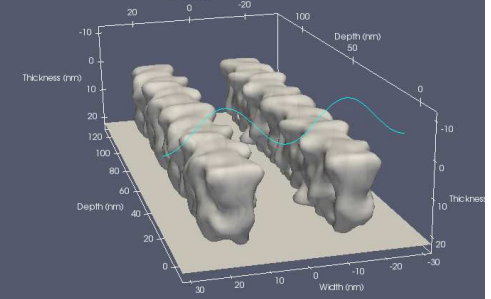


30 mJ/cm²

Map of Oxo-Networks

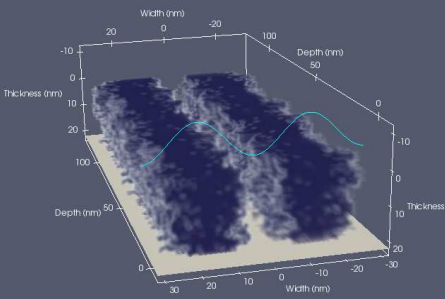


Developed Relief Image

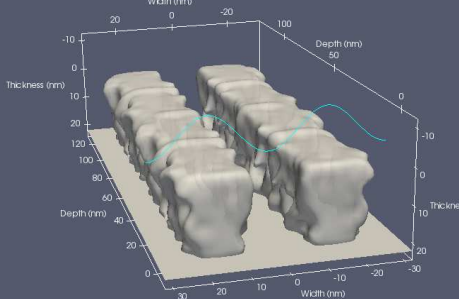


40 mJ/cm²

Map of Oxo-Networks

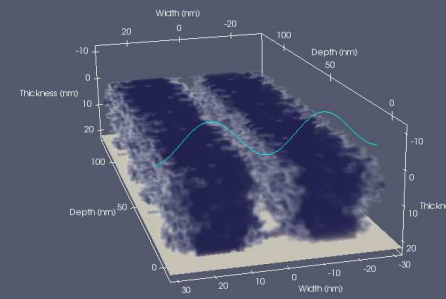


Developed Relief Image

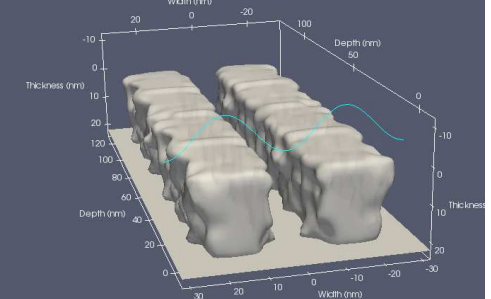


50 mJ/cm²

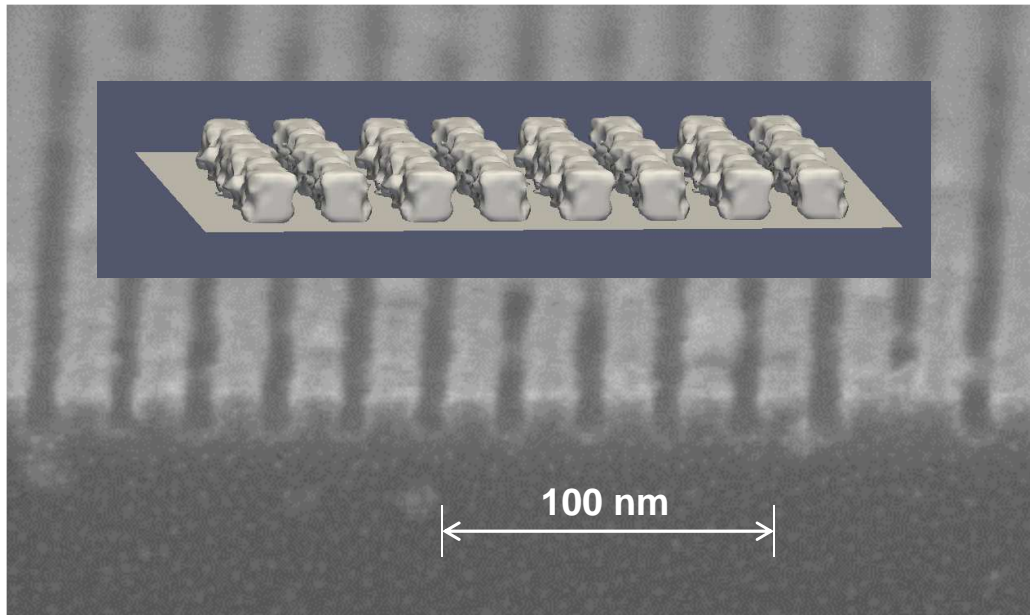
Map of Oxo-Networks



Developed Relief Image



Calculated vs Experimental Image : 24 nm pitch



Imaged using EUV-IL tool
(Paul Scherrer Institut)

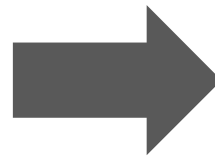
Summary

Simple chemical description of MOx resist

Photo-induced condensation of **multifunctional** cluster

EUV exposure chemistry data from **Inpria MOx resist** test vehicles

General **stochastic simulation** process



Quantitative link between photochemistry and imaging

Contrast originates from non-linear **oxo-network** formation

Lithographic predictions consistent with experimental observations

Potentially applicable to **many resist systems**